

## The Crystal Structure of $\text{Ti}_5\text{Te}_4$

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One of the phases in the titanium-tellurium system is identified as  $\text{Ti}_5\text{Te}_4$ . This compound is tetragonal with 10 Ti and 8 Te in the body-centered cell with dimensions

$$a = 10.164, \quad c = 3.7720 \text{ \AA}.$$

The space group is  $I4/m - C_{4h}^3$  and the atomic positions are:

$$\begin{aligned} &(0, 0, 0; \frac{1}{2}, \frac{1}{2}, \frac{1}{2}) +; \quad 2 \text{ Ti in } (a) \quad 0, 0, 0; \\ &8 \text{ Ti in } (h) \quad x, y, 0; \quad \bar{x}, \bar{y}, 0; \quad \bar{y}, x, 0; \quad y, \bar{x}, 0, \\ &\text{with } x_1 = 0.3144, \quad y_1 = 0.3752 \quad \text{and} \quad 8 \text{ Te in } (h) \\ &\text{with } x_2 = 0.0589, \quad y_2 = 0.2797. \end{aligned}$$

The relationships between the  $\text{Ti}_5\text{Te}_4$ -structure and the NiAs-like structure of the  $\text{Ti}_{2-x}\text{Te}_2$ -phase are discussed.

### Introduction

In a recent study of the titanium tellurides (Raaum, 1959) the existence of a phase with composition  $\text{Ti}_5\text{Te}_4$  was established. The composition was determined by means of X-ray and density measurements. Samples were prepared by melting weighed quantities of titanium and tellurium in the stoichiometric proportions at 1400–1500 °C. in crucibles of pure alumina, placed inside silica tubes which were evacuated and sealed. Single crystals were found in samples quenched from about 1200 °C. Guinier powder photographs could be indexed as tetragonal with the following cell dimensions:

$$a = 10.164, \quad c = 3.7720 \text{ \AA}; \quad c/a = 0.3711.$$

On the basis of the observed density, 6.343 g.cm.<sup>-3</sup>, the unit cell contains ten titanium and eight tellurium atoms.

The titanium tellurides have also been studied by Hahn & Ness (1957, 1959). They report *inter alia* the existence of a  $\text{Ti}_2\text{Te}$ -phase with composition in the range  $\text{TiTe}_{0.25}$  to  $\text{TiTe}_{0.7}$ . The tetragonal unit cell was found to have dimensions:

$$a = 14.37 \pm 0.01, \quad c = 3.590 \pm 0.005 \text{ \AA}; \quad c/a = 0.250$$

and the space group assumed to be  $P4/mmm - D_{4h}^1$ . The powder photograph data of Hahn & Ness leave little doubt that their  $\text{Ti}_2\text{Te}$ -phase is identical with the present  $\text{Ti}_5\text{Te}_4$ -phase. According to Raaum no extended range of homogeneity for this phase exists, and the composition is unequivocally determined to be  $\text{TiTe}_{0.80}$  or  $\text{Ti}_5\text{Te}_4$ .

The atomic arrangement in this structure is of

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considerable interest; especially its resemblance to the NiAs-like structure of the  $\text{Ti}_{2-x}\text{Te}_2$ -phase which also exists in the titanium tellurium system, and for which Raaum (1959) found  $0.39 \leq x \leq 1.00$ .

### Unit cell and space group

Well formed, needle-like, single crystals with almost quadratic cross-sections were investigated on the Weissenberg goniometer with Cu K radiation ( $\lambda_{\text{Cu}} = 1.5405 \text{ \AA}$ ). Weissenberg photographs were taken both along the needle axis and perpendicular to it.

The only systematic missing reflexions were of the type

$$hkl \quad \text{when} \quad h + k + l = 2n + 1.$$

The marked differences between  $I(hkl)$  and  $I(khl)$  show that the Laue symmetry is  $4/m$ . The highest symmetric space group is thus  $I4/m - C_{4h}^5$ .

### Determination of the structure

Intensity measurements of the  $hk0$  and  $hkl$ -reflexions were carried out visually on the Weissenberg photographs, using the multiple-film technique. Corrections for the combined Lorentz and polarization factor were made. For the calculation of the structure amplitudes the Thomas-Fermi-model values of  $f$  were taken from the *Internationale Tabellen*. Corrections for absorption and extinction were not applied. A temperature correction was attempted for the  $hk0$ -material, using individual temperature factors  $B$ .

The agreement between  $F_o$  and  $F_c$  is expressed by the reliability index  $R = \Sigma(|F_o| - |F_c|) / \Sigma|F_o|$ .  $F_c$ -values corresponding to undetected reflexions have been added with their half values.

Since the space group has the symmetry  $p4$  in the

Table 1. *Observed and calculated structure factors for Ti<sub>5</sub>Te<sub>4</sub>*

<i>hkl</i>	<i>F<sub>o</sub></i>	$\frac{1}{2}F_c$	<i>hkl</i>	<i>F<sub>o</sub></i>	$\frac{1}{2}F_c$	<i>hkl</i>	<i>F<sub>o</sub></i>	$\frac{1}{2}F_c$	<i>hkl</i>	<i>F<sub>o</sub></i>	$\frac{1}{2}F_c$
200	18.8	-27.1	950	69.7	+67.7	1,13,0	48.4	-52.1	651	153.5	+158.0
400	42.7	+51.0	11,5,0	40.0	+40.8				851	29.5	-24.8
600	42.8	-42.3				101	28.5	+46.0	10,5,1	62.3	+59.3
800	48.8	-45.0	260	82.7	+85.9	301	138.8	+149.3			
10,0,0	0	-8.0	460	33.0	+33.0	501	78.8	-71.5	161	64.5	-61.5
12,0,0	59.0	-49.3	660	39.7	+41.7	701	16.5	+10.0	361	17.0	-20.3
			860	101.6	+98.1	901	104.3	-106.3	561	37.0	-41.0
110	8.0	+11.2	10,6,0	51.1	-45.7	11,0,1	13.5	+18.5	761	21.0	-18.0
310	35.4	-52.3							961	56.8	+59.0
510	70.3	-93.5	170	142.3	+154.0	211	20.5	+29.5			
710	11.3	+11.8	370	0	-4.0	411	146.3	+150.5	271	123.8	+148.8
910	0	+0.2	570	59.5	+65.6	611	27.0	+21.0	471	69.0	-77.8
11,1,0	133.2	+121.6	770	82.4	-88.0	811	13.5	+19.5	671	0	-7.3
			970	15.0	-10.8	10,1,1	40.0	+41.8	871	58.0	-53.8
220	85.0	-106.0	11,7,0	27.3	-25.3	12,1,1	72.5	-82.5	10,7,1	10.0	-13.0
420	111.0	+167.4									
620	17.3	-26.0	280	18.8	+14.0	121	106.8	-135.3	181	50.0	+44.8
820	95.6	+96.4	480	61.2	-62.4	321	39.5	+41.3	381	44.8	+48.3
10,2,0	44.9	+46.9	680	0	-8.5	521	84.7	-79.0	581	73.3	+66.0
12,2,0	0	+1.6	880	34.4	-36.5	721	87.8	+82.8	781	0	+0.7
			10,8,0	19.5	+18.4	921	65.0	+63.3	981	60.0	+64.0
130	66.8	+95.4				11,2,1	69.0	+65.0			
330	89.1	+87.1	190	68.7	-67.7				291	0	-2.2
530	86.6	+88.7	390	47.3	-44.0	231	38.0	-46.3	491	32.3	-29.0
730	8.0	+3.5	590	97.3	+101.7	431	69.5	-71.2	691	43.5	+46.3
930	60.4	-57.7	790	0	-1.1	631	0	-3.5	891	74.3	+77.7
11,3,0	13.0	-8.6	990	94.0	+88.8	831	79.8	-69.8			
						10,3,1	85.5	+72.8	1,10,1	0	+6.8
240	15.0	-23.5	2,10,0	80.9	+79.9	12,3,1	30.5	-31.0	3,10,1	71.5	-71.5
440	21.2	+25.0	4,10,0	8.0	15.8				5,10,1	30.3	+32.7
640	106.4	-107.1	6,10,0	80.9	+82.5	141	0	-2.5	7,10,1	69.3	-86.5
840	36.2	+38.9	8,10,0	18.9	-23.3	341	77.5	+101.5			
10,4,0	77.7	-82.2				541	34.5	+34.5	2,11,1	56.7	+57.7
12,4,0	70.2	+61.9	1,11,0	0	+7.7	741	25.3	+23.8	4,11,1	76.8	+66.8
			3,11,0	38.4	+33.2	941	60.5	-71.0	6,11,1	23.0	-25.0
150	30.5	+36.5	5,11,0	48.4	-44.7	11,4,1	58.0	-55.0			
350	129.5	-133.8	7,11,0	51.7	-55.1				1,12,1	51.0	+57.0
550	9.8	+11.3				251	25.5	+28.0	3,12,1	17.0	-19.5
750	33.1	+39.9	2,12,0	13.9	+17.6	451	0	-7.0			
			4,12,0	76.9	-72.7						

001-projection, with  $a = \frac{1}{2}\sqrt{(2a)} = 7.1871 \text{ \AA}$ , it was decided to start with a Patterson projection on the basis of the corrected  $F^2(hk0)$ -values. The 001-direction is also specially suitable because of the short  $c$ -axis and because the crystal has a section not far from circular along the  $c$ -axis.

The primitive projection contains five titanium and four tellurium atoms; one titanium atom must thus be located either in 0, 0 or  $\frac{1}{2}, \frac{1}{2}$ .

The Patterson map showed the highest peak at  $u = 0.444$  and  $v = 0.117$ . This peak, and a peak at  $u = 0.444$ ,  $v = 0.336$  with about 40% of the value for the highest peak, are probably due to vectors between the tellurium atoms. Two peaks at  $u = 0.056$ ,  $v = 0.278$  and  $u = 0.338$ ,  $v = 0.224$  with about 40% and one peak at  $u = 0.168$ ,  $v = 0.348$  with about 28% of the value of the highest peak are also present in the Patterson map. They are probably due to vectors between titanium and tellurium atoms.

A set of signs was calculated on the basis of the preliminary parameters for the tellurium atom and the titanium atom in the (0, 0) position. An electron-density projection was evaluated, and this clearly

showed all atoms well resolved with a distinctly higher electron density in the assumed tellurium position than in the other atomic positions.

The atomic parameters were refined further by means of two Fourier and two difference syntheses.  $R$  decreased to 0.18 after the third Fourier synthesis and to 0.112 after the second difference synthesis. The observed and calculated  $F(hk0)$ -values are listed in Table 1, and the final Fourier map is shown in Fig. 1. The final parameters, referred to the body-centered axes, are:

$$\begin{aligned} &2 \text{ Ti I in } 0, 0, \text{ etc.} \\ &8 \text{ Ti II in } x_1 = 0.3144, y_1 = 0.3752, \text{ etc.} \\ &8 \text{ Te in } x_2 = 0.0589, y_2 = 0.2797, \text{ etc.} \end{aligned}$$

In order to determine the effect of the thermal vibrations on the final parameters of the atoms, as derived from the difference synthesis, individual  $B$ -values were computed by the method of least-squares on the Ferranti Mercury computer ('Frederic') at the Norwegian Defence Research Establishment. The refinement was terminated after five cycles. The shifts in atomic coordinates, and the  $B$ -values were:

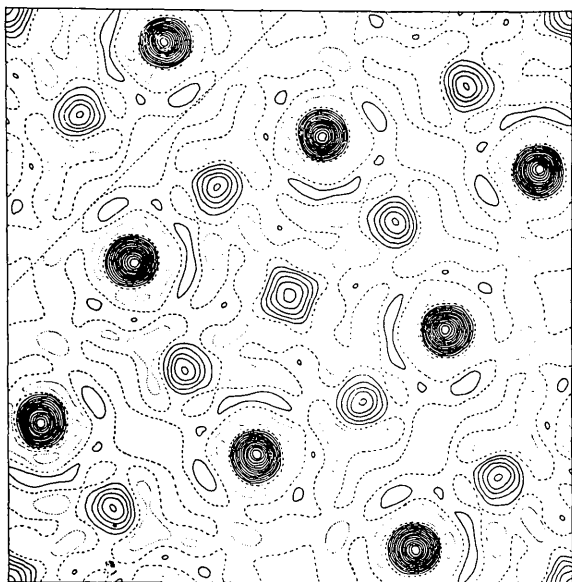


Fig. 1. Fourier projection of  $Ti_5Te_4$  on (001). Contours are at intervals of  $10 \text{ e.}\text{\AA}^{-2}$ . The zero contours are broken and the negative contours dotted.

$$\begin{array}{l} \text{Ti I } \Delta x = 0, \quad \Delta y = 0, \quad B = -1.019 \\ \text{Ti II } \Delta x_1 = -0.0001, \quad \Delta y_1 = +0.0005, \quad B = -0.892 \\ \text{Te } \Delta x_2 = -0.0005, \quad \Delta y_2 = +0.0001, \quad B = -0.799 \end{array}$$

and at this stage  $R$  was 0.092. The apparently negative  $B$ -values most probably represent absorption; they became 0.7–0.9 after correction for absorption ( $\mu R = 2.5$ ).

The short  $c$ -axis indicates that the  $z$ -parameter for all atoms are either 0 or  $\frac{1}{2}$ . However, the crystals were needle-shaped in the direction of the  $c$ -axis, and the intensities for other projections than (001) were therefore subject to absorption errors which are very difficult to correct for; for this reason, no success was obtained with these projections. The only possibility left was to evaluate intensities of higher layer lines with the  $c$ -axis as rotation axis. The assumed  $z$ -coordinates were tested by comparing  $|F(hk1)|_o$  with  $|F(hk1)|_c$ . The agreement was so good,  $R = 0.114$ , that the  $z$ -coordinates selected must be correct (see Table 1).

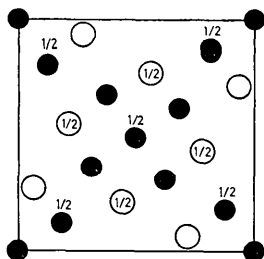


Fig. 2. The structure of  $Ti_5Te_4$  viewed along the  $c$ -axis. Filled circles represent titanium atoms and open circles represent tellurium atoms. The  $z$ -parameters of the atoms are either 0 or  $\frac{1}{2}$  as indicated.

The deduced atomic arrangement (see Fig. 2) is thus

$$\begin{array}{l} (0, 0, 0; \frac{1}{2}, \frac{1}{2}, \frac{1}{2}) + \\ 2 \text{ Ti I in } (a) 0, 0, 0. \\ 8 \text{ Ti II in } (h) x, y, 0; \bar{x}, \bar{y}, 0; \bar{y}, x, 0; y, \bar{x}, 0, \\ \quad \text{with } x_1 = 0.3144, y_1 = 0.3752. \\ 8 \text{ Te in } (h) \text{ with } x_2 = 0.0589, y_2 = 0.2797. \end{array}$$

The interatomic distances between nearest neighbours calculated using these parameters are listed in Table 2.

Table 2. *Interatomic distances in  $Ti_5Te_4$*

Ti I–2 Ti I	: 3.772 Å
–8 Ti II	: 2.954 ± 0.009 Å
–4 Te	: 2.905 ± 0.002
Ti II–2 Ti I	: 2.954 ± 0.009 Å
–2 Ti II	: 3.215 ± 0.019
–2 Ti II	: 3.428 ± 0.019
–2 Ti II	: 3.772
–1 Te	: 2.773 ± 0.010
–2 Te	: 2.774 ± 0.010
–2 Te	: 2.821 ± 0.010
Te–1 Ti I	: 2.905 ± 0.002 Å
–1 Ti II	: 2.773 ± 0.010
–2 Ti II	: 2.774 ± 0.010
–2 Ti II	: 2.821 ± 0.010
–2 Te	: 3.772
–4 Te	: 3.782 ± 0.003
–2 Te	: 4.110 ± 0.003
–2 Te	: 4.359 ± 0.003

The limits indicated for the interatomic distances are standard deviations calculated according to the method of Cruickshank (1949).

### Discussion of the structure

The coordination around the titanium and tellurium atoms can be seen from Fig. 3, where two unit cells of the  $Ti_5Te_4$  structure are shown in projection on (001). The structure might be looked upon as built of quadratic columns of titanium and tellurium atoms in the direction of the  $c$ -axis. Each of the two titanium atoms (Ti I) in position (a) is surrounded by four tellurium atoms in a planar quadratic arrangement, and by eight titanium atoms (Ti II) in a prismatic, approximately cubic, arrangement. Each Ti II atom is surrounded by five tellurium atoms, four of them forming a rectangular arrangement. It is furthermore surrounded by two Ti I atoms and by six Ti II atoms, of which two are on one side and four on the other side of the plane formed by the four tellurium atoms. Each tellurium atom is located at one apex of a rectangular double pyramid formed by a rectangle of four Ti II atoms and one Ti I atom at the other apex. A sixth titanium neighbour (Ti II) is found in an adjacent column. Each tellurium atom is furthermore surrounded by ten tellurium atoms at distances shorter than 4.36 Å.

The phase  $Ti_5Te_4$  has an arrangement of atoms typical of the metallic state and correspondingly short

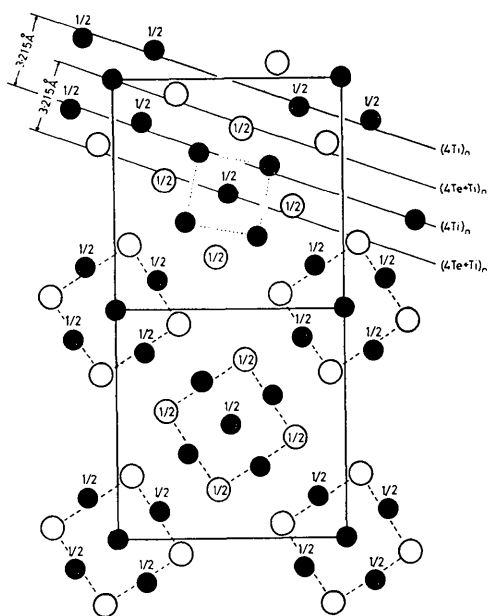


Fig. 3. Two unit cells of  $Ti_5Te_4$  viewed along the  $c$ -axis. The configuration of tellurium atoms around the titanium atoms is indicated by broken lines. Dotted lines indicate the relationships to the  $\beta$ -Ti structure. As can be seen from the upper part of the figure, the atoms are arranged in layers approximately parallel to (310).

metal-metal and metal-non-metal distances. The titanium-tellurium distances are even shorter than the sum of the metallic radii given by Pauling (1947):  $1.467 + 1.51 = 2.977$  Å for the twelve-coordinated titanium I and six-coordinated tellurium atoms, and  $1.397 + 1.51 = 2.907$  Å for the seven-coordinated titanium II and six-coordinated tellurium atoms, respectively. The titanium coordination around the Ti I atoms is the same as in the cubic body-centered  $\beta$ -titanium structure. According to Levinger (1953) the cube edge of  $\beta$ -titanium is 3.283 Å and the shortest interatomic distances 2.843 Å. In  $Ti_5Te_4$  the edges in the direction of the  $a$ - and  $c$ -axis are 3.215 and 3.772 Å, respectively, with the shortest interatomic distances being 2.954 Å. The titanium coordination around the Ti II atoms is more irregular, and the distances to the eight nearest neighbours range from 2.954 to 3.772 Å.

The structure of the  $Ti_5Te_4$ -phase shows characteristic relationships to that of the neighbouring  $Ti_{2-x}Te_2$  phase with NiAs-like structure in the two-component system. According to Raaum (1959) the titanium-rich limit of the phase lies at  $Ti_{1.61}Te_2$ , and the structure is not hexagonal as assumed by Ehrlich (1949), but monoclinic with unit cell dimensions:

$$a = 6.954, b = 3.836, c = 12.716 \text{ \AA}; \beta = 90.63^\circ.$$

In Fig. 4 two unit cells of  $Ti_{1.61}Te_2$  are projected into the  $ac$ -plane, assuming the parameters of an ideal NiAs-like structure since the actual parameters are not yet known. It should also be remarked that about

20% of the metal sites are unoccupied, probably in an ordered way as indicated by the crosses, by analogy with what has been found for the structurally related compounds  $Ni_{0.80}Se$  (Grønvold, 1955),  $Fe_3Se_4$  (Okazaki & Hirakawa, 1956) and  $Cr_3S_4$  (Jellinek, 1957).

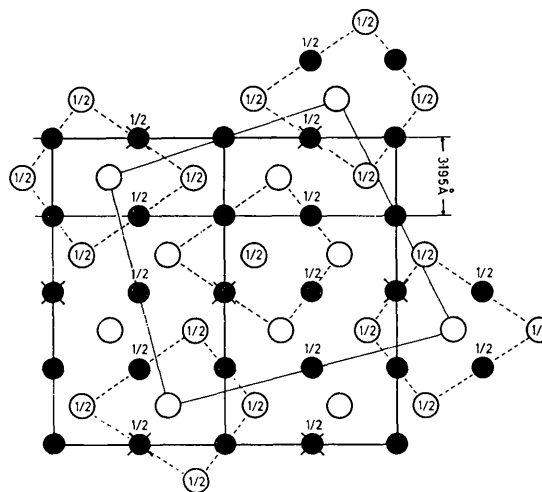


Fig. 4. Two unit cells of a NiAs-type structure with the dimensions of the  $Ti_{1.61}Te_2$ -structure. Probable locations of the vacancies are indicated by crosses.

By comparing Figs. 3 and 4 it is seen that the irregular quadrangle outlined in Fig. 4, with edge-lengths of 9.9, 10.6, 9.8 and 12.0 Å for the  $Ti_{2-x}Te_2$ -phase, has its characteristic counterpart in the square plane cell with  $a = 10.164$  Å for the  $Ti_5Te_4$ -phase. The small dotted quadrilaterals in Fig. 4 resemble those in Fig. 3 except for the fact that the central atoms are tellurium in Fig. 4 and titanium atoms in Fig. 3, and that although the four titanium atoms in both cases show a quadratic arrangement, two of them are at different heights in the  $Ti_{2-x}Te_2$ -structure. Inside the unit formed by the large quadrilateral in Fig. 4 there are ten tellurium atoms and eight titanium atoms when the composition of the phase is  $Ti_{1.60}Te_2$ . In order to obtain the  $Ti_5Te_4$ -structure from that of  $Ti_{1.61}Te_2$  it is thus only necessary to substitute the tellurium atoms at the centers of the dashed quadrilaterals by titanium atoms and rearrange the vacancies so that they are now on the sides of the large quadrilateral. Evidently, these two possibilities are energetically favoured over one resulting in a compound with composition  $TiTe$  and NiAs-type structure.

Compounds structurally related to  $Ti_5Te_4$  seem to exist in the vanadium-sulfur system (Pedersen, 1958) and in the vanadium-tellurium system (Grønvold *et al.*, 1958).  $V_5S_4$  is probably isostructural with  $Ti_5Te_4$ , while  $V_5Te_4$  has a monoclinic structure of similar dimensions.

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## Experimental Atomic Scattering Factors and Anomalous Dispersion Corrections for Th, U, and Pu\*†

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Experimental atomic scattering factors have been determined for Th, U, and Pu with Mo  $K\alpha$ , Cu  $K\alpha$ , Fe  $K\alpha$ , and Cr  $K\alpha$  X-radiations. The Thomas-Fermi-Dirac scattering curves were used as a theoretical basis, and the difference between the experimental and TFD curves was taken as a measure of the anomalous dispersion correction.

As a result of determining the scattering factors for Th, U, and Pu from experimental samples of  $ThO_2$ ,  $UO_2$ , and  $PuO_2$ , the scattering factor for oxygen was also determined. The experimentally derived scattering curve for oxygen is in good agreement with the theoretical scattering curve for oxygen according to McWeeney.

Calculated values for  $\Delta f'$  and  $\Delta f''$ , the real and imaginary portions of the anomalous dispersion correction are compared with experimental values for these quantities. Agreement can be described as semi-quantitative since the experimental terms are of the same order of magnitude and have the same signs as those indicated by theory.

### Introduction

Accurate atomic scattering factors are essential for any careful study of interatomic distances and thermal vibration parameters by diffraction methods. Because of the interest in this laboratory in precise studies of the structures of intermetallic and other compounds of various heavy metals, the atomic scattering factors of Th, U, and Pu have been determined experimentally for Mo, Cu, Fe, and Cr  $K\alpha$  X-radiations.

The Thomas-Fermi-Dirac scattering curves are used as a theoretical basis, and the difference between the experimental curves and the TFD curves is taken as a measure of the anomalous dispersion correction. The  $L_{III}$  absorption edges of Th, U, and Pu lie very close to Mo  $K\alpha$  radiation, and the  $M_I$ ,  $M_{II}$  absorption edges of these elements are close to Fe  $K\alpha$  and Cr  $K\alpha$

radiation respectively. Since no really sound theoretical treatment of anomalous dispersion for the  $L$  and  $M$  electrons has been developed, the agreement between the experimental and calculated values for anomalous dispersion in this region must be described as qualitative or at best semi-quantitative. Furthermore, any theoretical treatment is complicated because the  $L$  and  $M$  absorption edges themselves are quite complex.

### Experimental theory

The atomic scattering factors of Th, U, and Pu were experimentally determined by utilizing techniques first proposed by the author (Roof, 1959a), corrected by Chipman and Paskin (1960), and subsequently revised by the author (Roof, 1960a). A brief description of these experimental techniques is given in the following paragraphs.

The experimental samples consisted of a powdered material having the geometrical shape of a slab. The equation relating the atomic scattering factor of a material having the shape of a slab to experimental

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† A brief summary of this paper was presented at the Fifth Congress of the International Union of Crystallography at Cambridge, England, August, 1960.